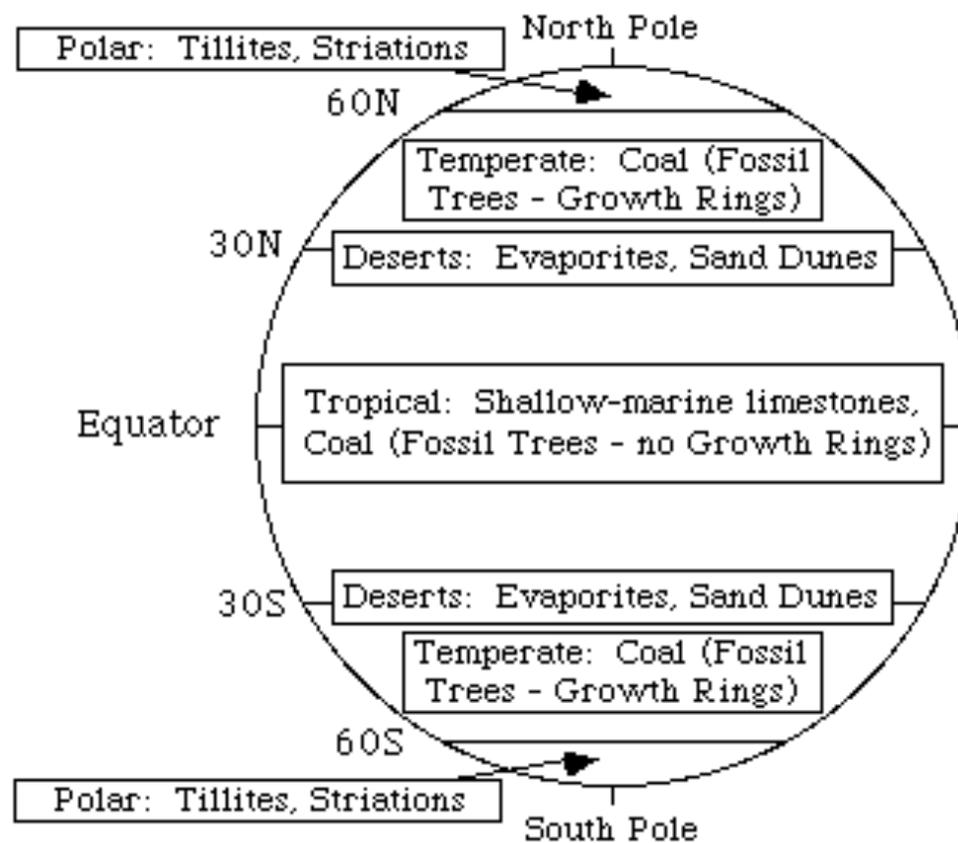
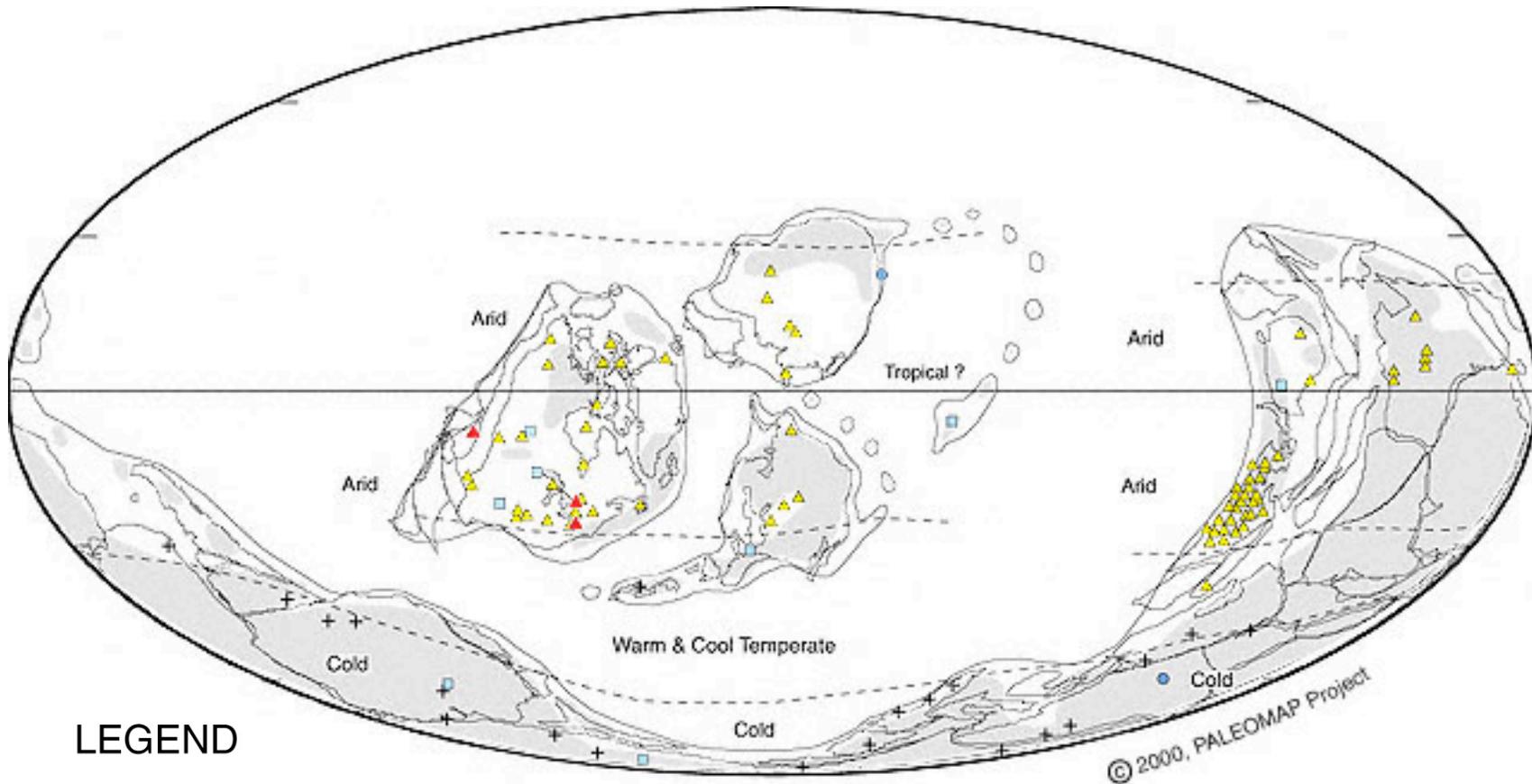


# Paleoclimate indicators

# Rock types as indicators of climate



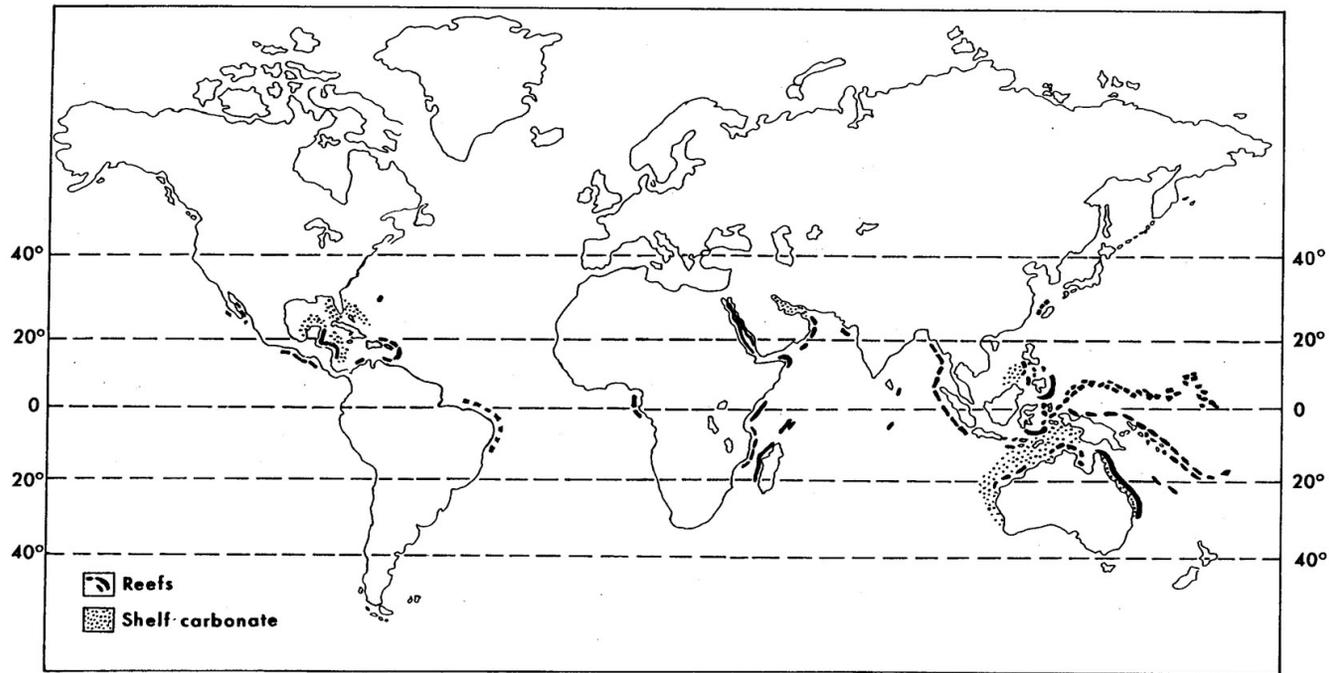


### LEGEND

		WARM	COOL
		<i>Tropical</i>	<i>Cool Temperate</i>
WET		● Coal	● Coal & Tillites
		● Bauxite	
		● Laterite	
		<i>Warm Temperate</i>	
		■ Kaolinite (& coal & evaporite)	
		🌴 Crocodiles	🌴 Palms & Mangroves
DRY		▲ Evaporite	⊕ Tillite
		▲ Calcrete	⊕ Dropstone
			● Glendonite
		<i>Arid</i>	<i>Cold</i>

## Middle & Upper Ordovician

"Paratropical" = High Latitude Bauxites



. Distribution of modern marine carbonate sediments in shallow water

Accumulation of significant thicknesses of limestone and reef-bearing limestone is restricted to  $\sim 20^\circ \pm$  equator

# Gowganda tillite, Ontario

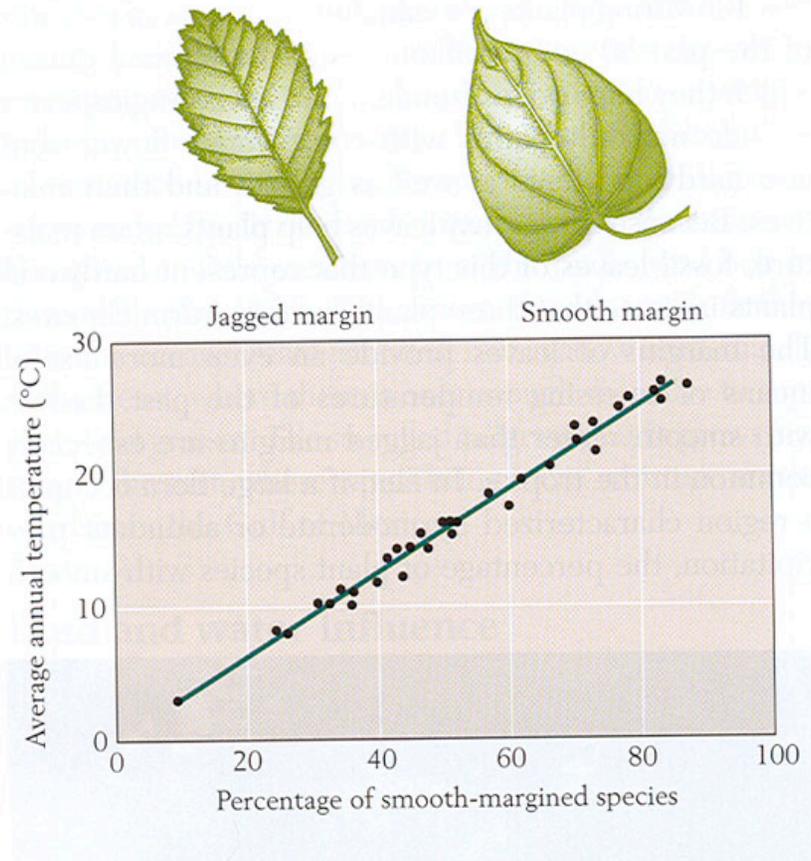


Rock record of glaciation, this one ~2.3 billion years ago!

## The concept of climate proxies

- A climate proxy is something that records or reflects a change in temp or rainfall but does not **DIRECTLY** measure temperature or precipitation

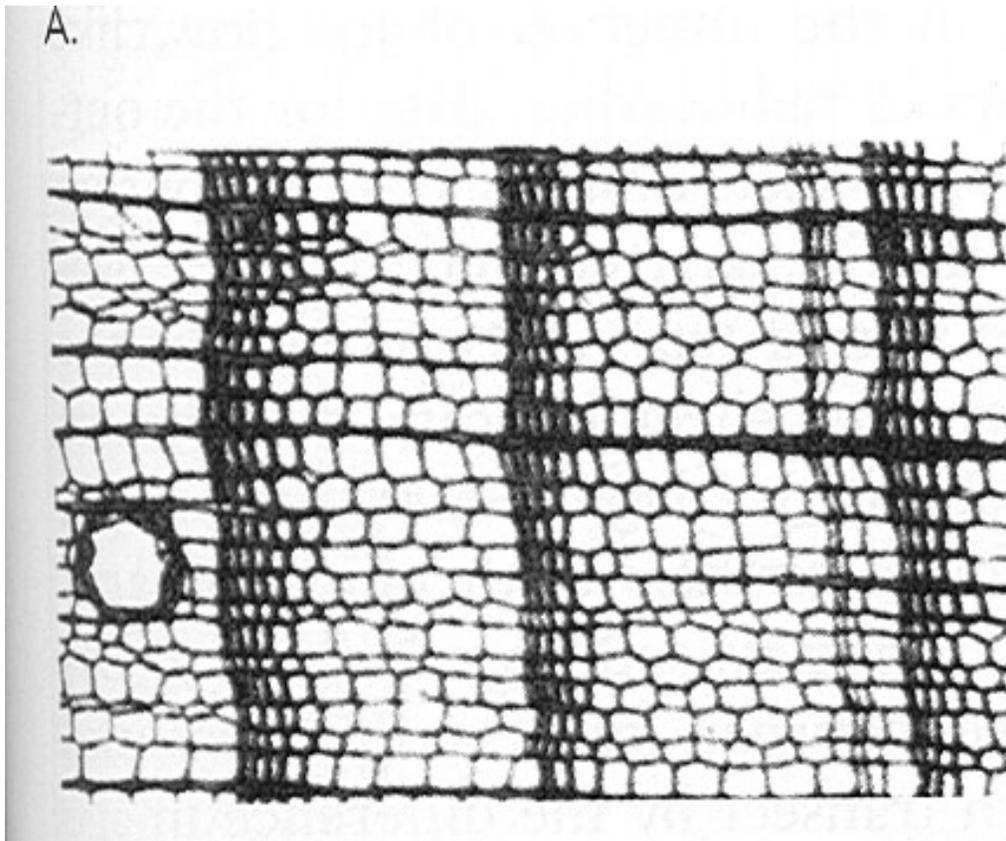
For example...



Serrations on leaf margins indicate temperature: jagged edges indicate cooler climate

Leaf waxiness is an indicator of moisture retention

# Tree rings:degree of seasonality



In the tropics there is no difference in growing seasons and tree rings are not well developed. These are tree rings from a tree in a temperate climate

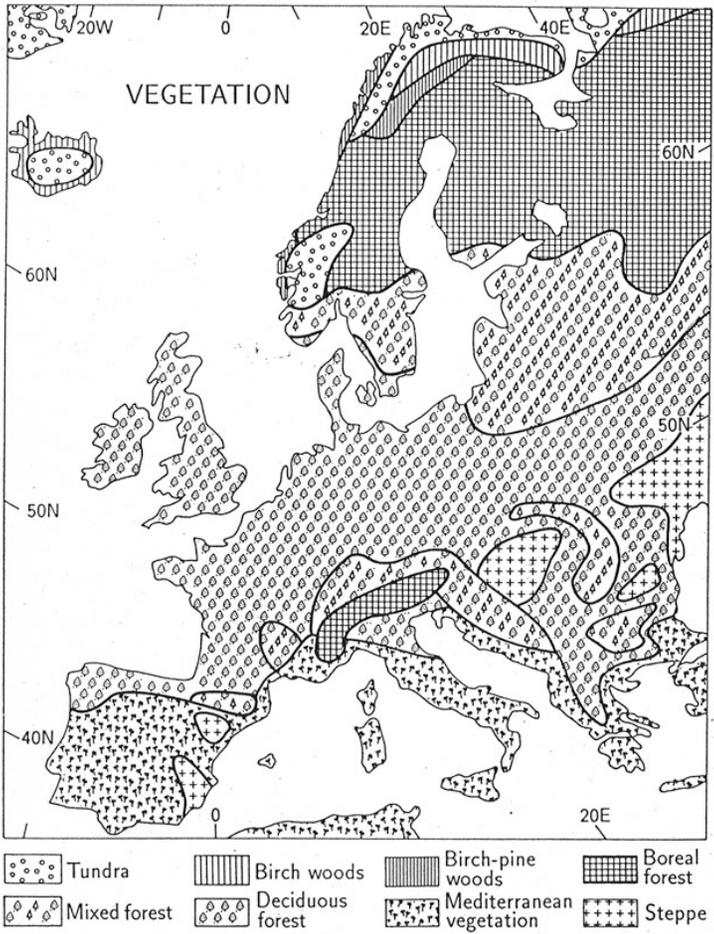
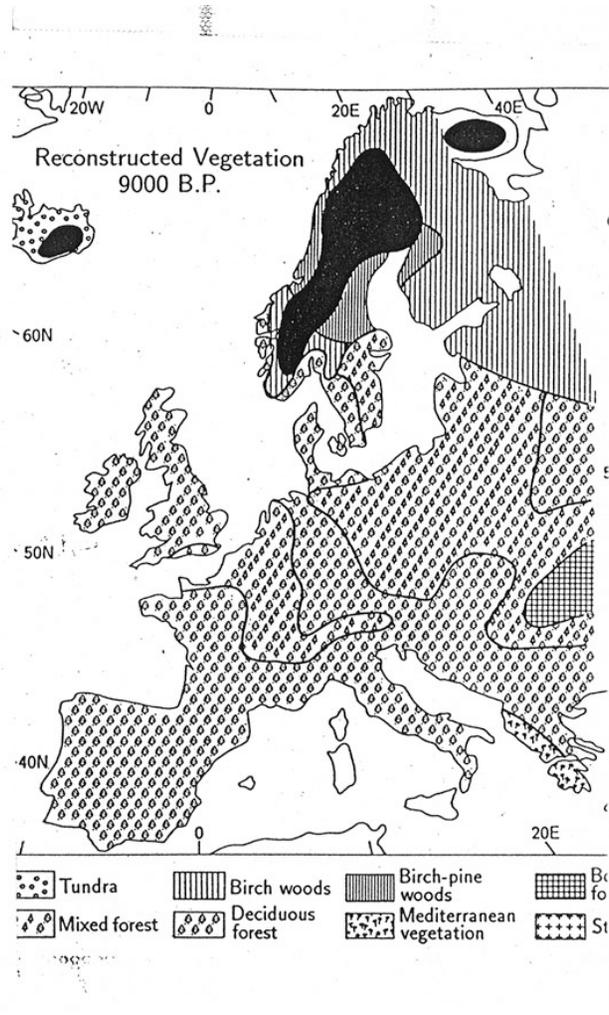
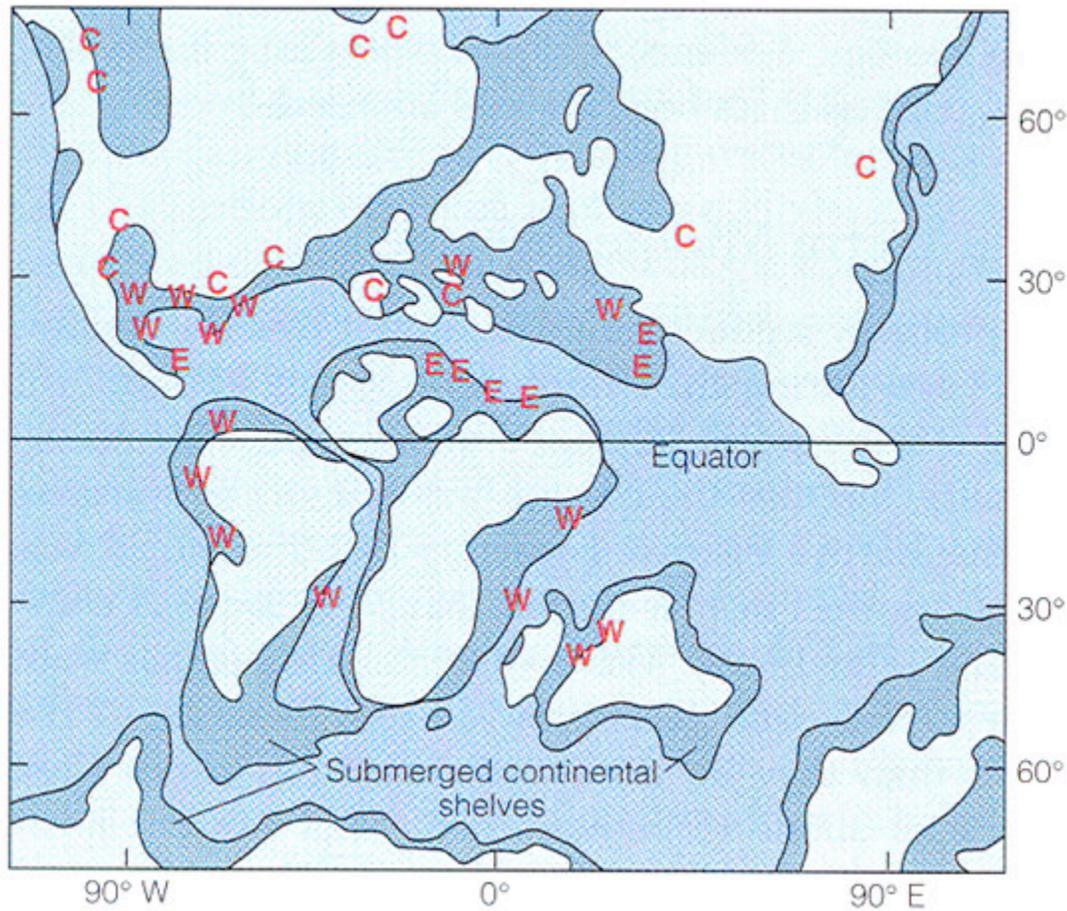


Fig 73. European vegetation regions (data from Fullard and



Vegetation zones in Europe parallel latitude now and 9,000 yr before present

# Middle Cretaceous climate indicators



W = warm-water animal occurrences;  
E = evaporite minerals;  
C = coal deposits. Coal is not particularly temp sensitive but animals and evaporation are.

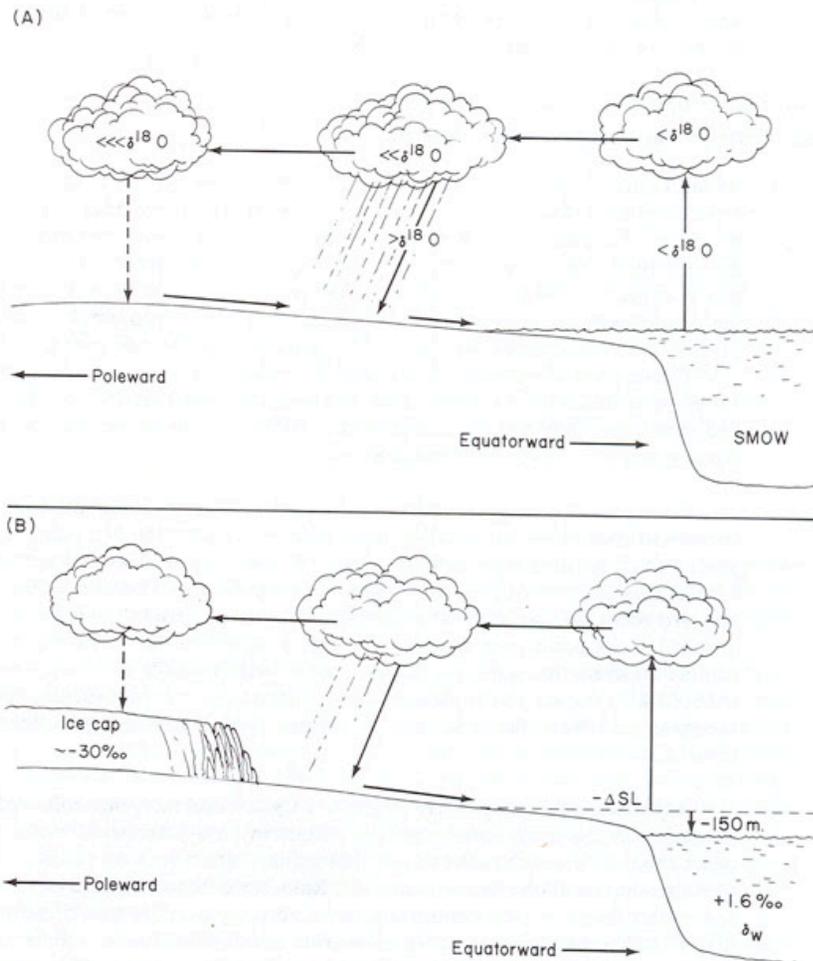
# Stable isotopes of oxygen: a proxy for temperature but now considered an almost direct measurement

- Stable isotopes: do not decay over time. Ex,  $O_{16}$  and  $O_{18}$ .

$O_{18}$  is produced from  $O_{16}$  through nucleosynthesis in supernovae

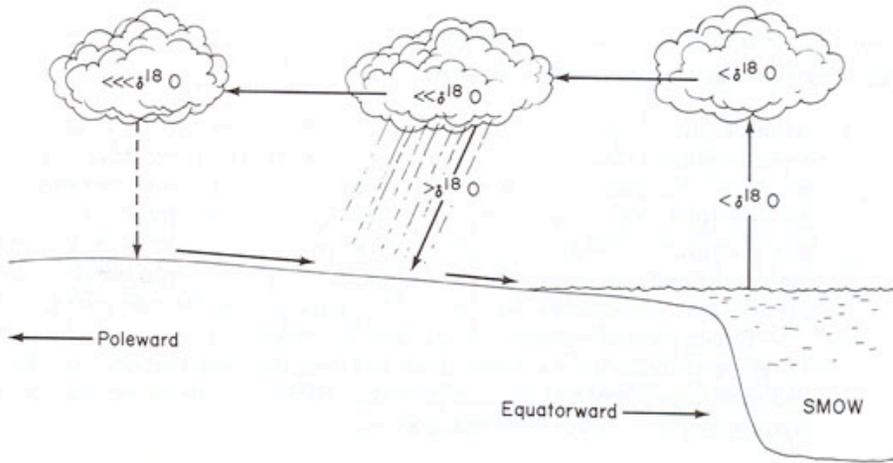
Anything that incorporates oxygen into its chemical structure will do so with some ratio of  $O_{18}:O_{16}$ . We can measure the ratios of these isotopes in the lab

# $O_{18}/O_{16}$ fractionation

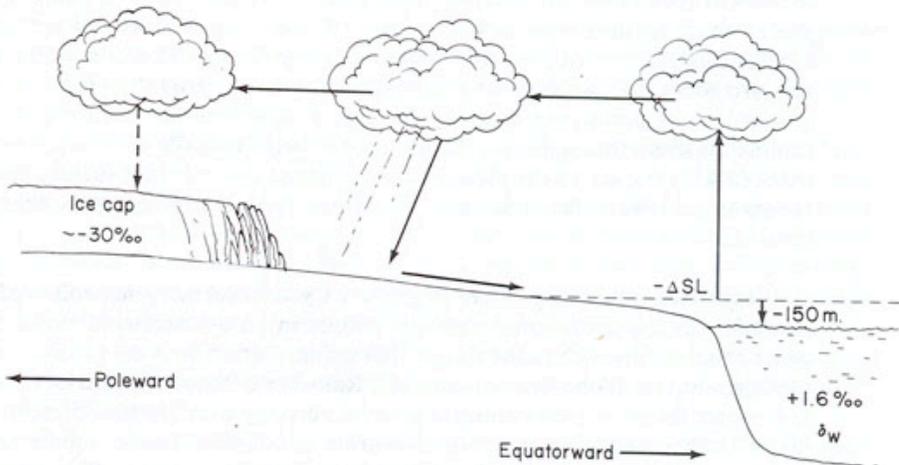


$H_2O$  is evaporated from sea water. The oxygen in the  $H_2O$  is enriched in the lighter  $O_{16}$ . This  $H_2O$  condenses in clouds, falling on land as precipitation. Thus,  $H_2O$  that is part of the terrestrial water cycle is enriched in the light  $O_{16}$  isotope and sea water is enriched in the heavier  $O_{18}$  isotope.

(A)



(B)



Glacial ice is therefore made up primarily of water with the light  $O_{16}$  isotope. This leaves the oceans enriched in the heavier  $O_{18}$ , or “more positive.” During glacial periods, more  $O_{16}$  is trapped in glacial ice and the oceans become even more enriched in  $O_{18}$ . During interglacial periods,  $O_{16}$  melts out of ice and the oceans become less  $O_{18}$  rich, or “more negative” in  $O_{18}$ .

If we collect a shell made out of  $\text{CaCO}_3$ , we can analyze the  $\text{O}_{18}:\text{O}_{16}$  ratio by the following formula:

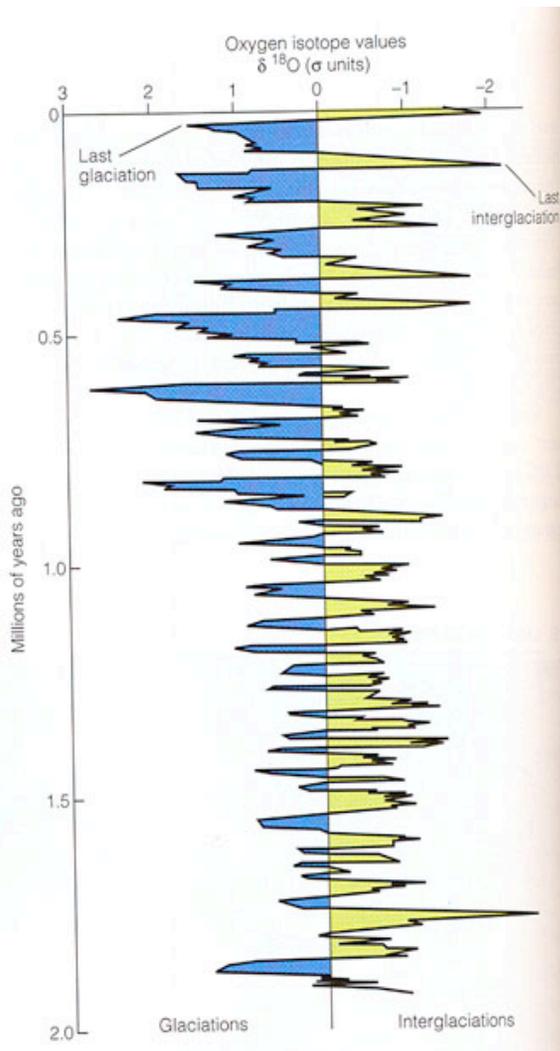


$$\delta \text{O}_{18} = \left[ \left( \frac{\text{O}_{18}/\text{O}_{16} \text{ sample}}{\text{O}_{18}/\text{O}_{16} \text{ standard}} \right) - 1 \right] \times 1000$$

The standard that your sample is compared to is either one prepared from ocean water or from a fossil standard. Positive  $\delta \text{O}_{18}$  values mean that your sample is enriched in the heavy O isotope; negative  $\delta \text{O}_{18}$  values mean it's depleted in the heavy  $\text{O}_{18}$ .



# Curve of average $O_{18}$ isotope variation over the past 2my based on analysis of deep sea sediment

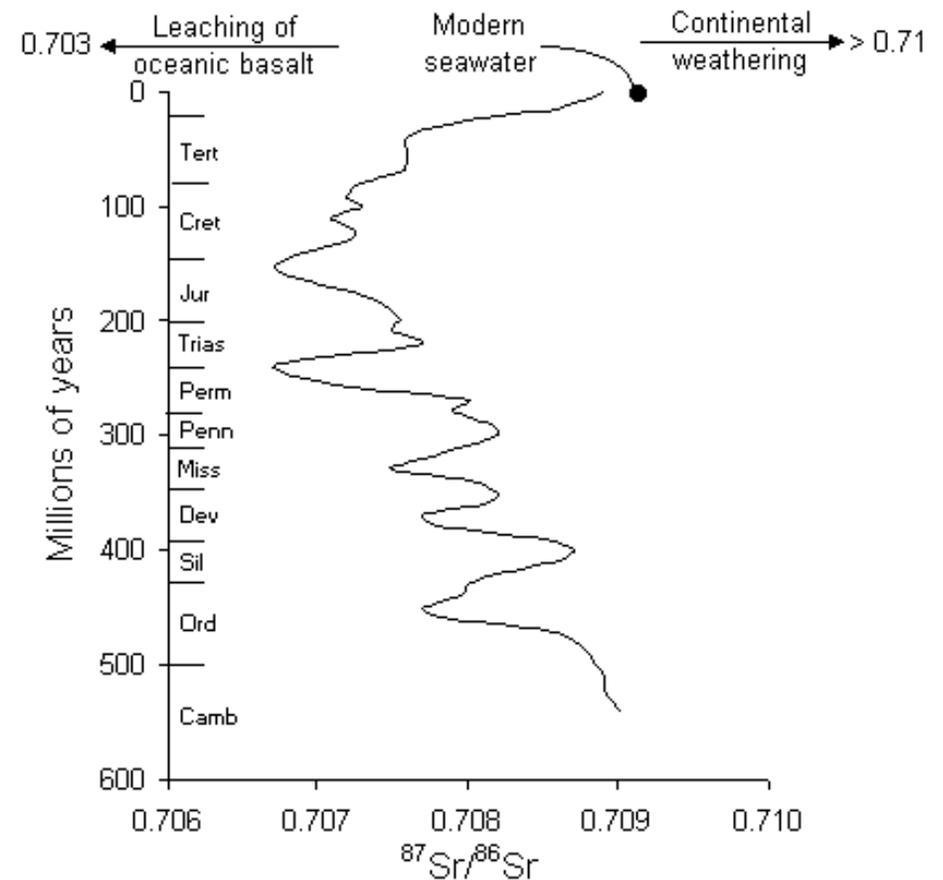
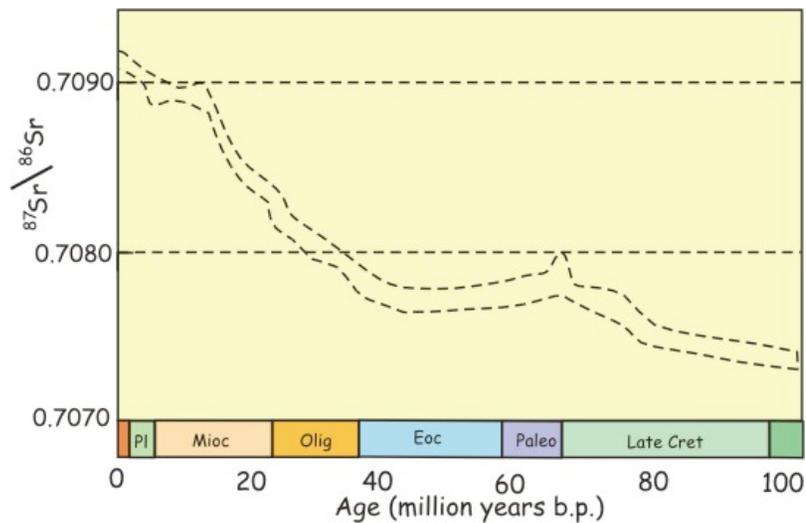


The curve illustrates changes in global ice volume in successive glacial (blue) and interglacial (green) cycles of the Quaternary Period.

Note that this graph does not show fluctuating temperatures, but changes in the dilution of sea water as a result of freshwater influx from melting glacial ice, i.e., this is a climate proxy

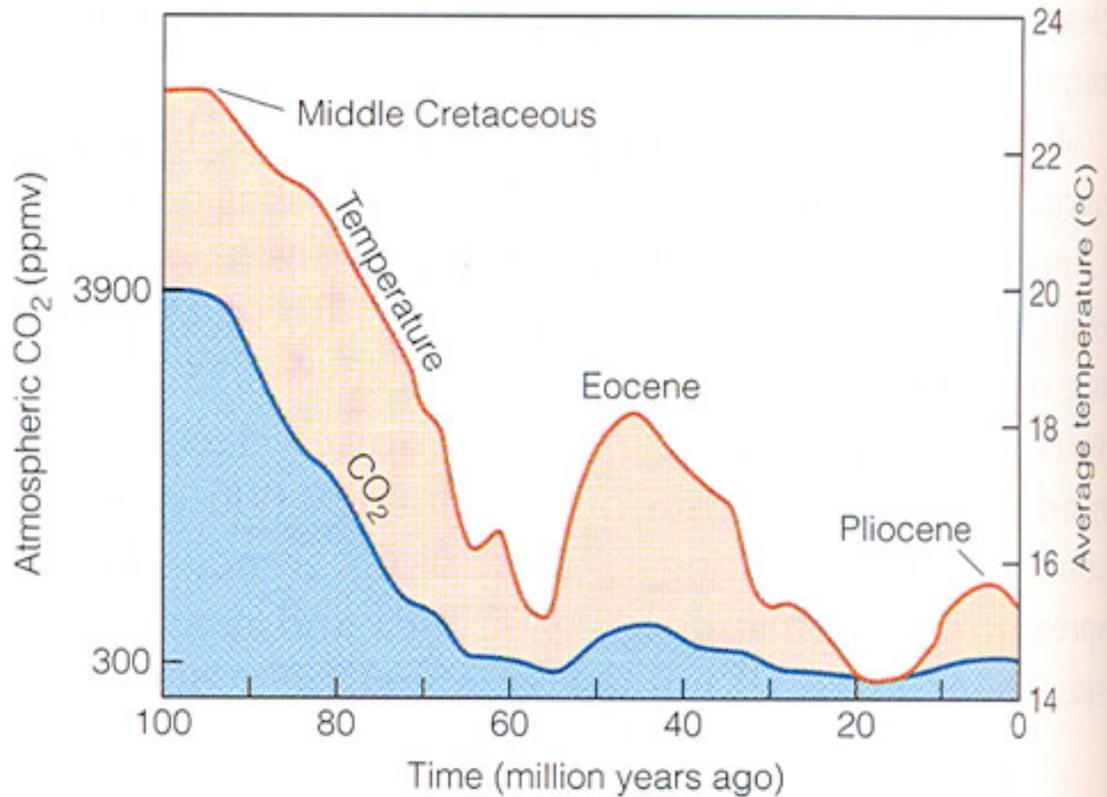
$$\text{Sr}_{87}/\text{Sr}_{86}$$

a proxy for salinity/rainfall (evap vs ppt)



During wet climates, more Sr is released from weathering minerals in granite

# Middle Cretaceous and early Tertiary temperatures



A reconstruction of changing atmospheric CO<sub>2</sub> levels (blue) and resulting global temps (pink) over the past 100 my, based on analysis of climate proxies such as pollen, leaf serration, and O<sub>18</sub> values. Note that the Mid-Late Cretaceous was significantly warmer than the early Tertiary (or today)

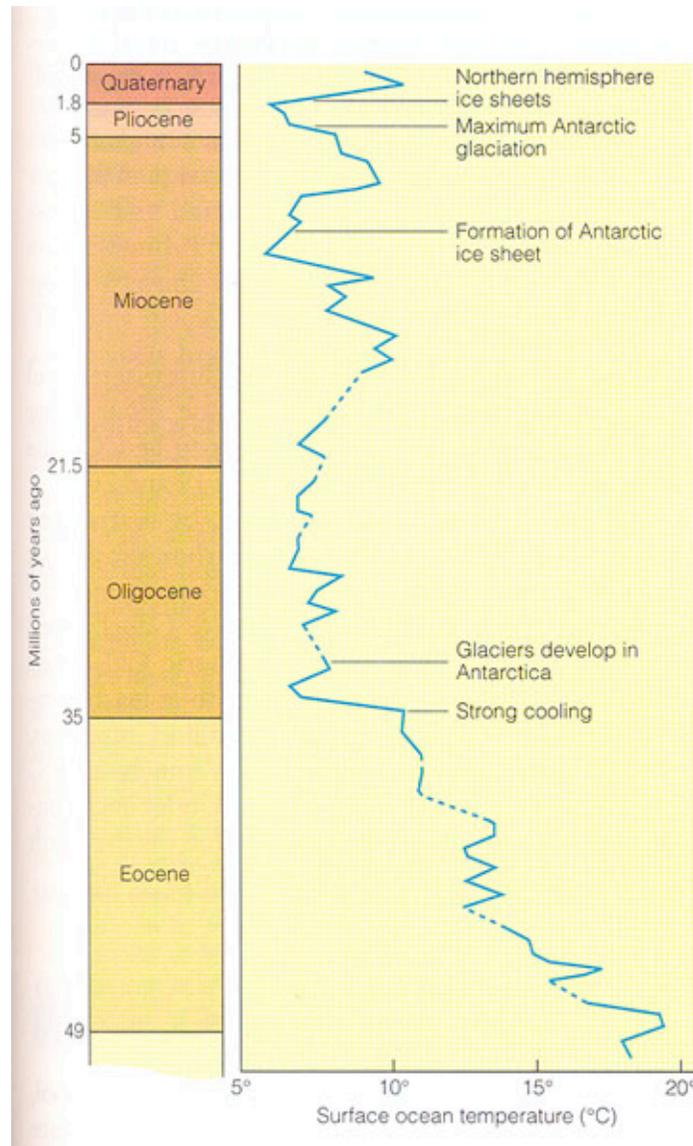
# Not proxies, but direct measures Sr:Ca ratios in aragonite: a measure of seawater temperature

Trace elements such as cadmium, barium, manganese and strontium replace Ca in the lattice of aragonite (a  $\text{CaCO}_3$  mineral). These trace elements often substitute for Ca as a result of climate-related variables.

The following relationship between Sr:Ca ratios and temp was developed:  $\text{Sr:Ca} \times 10^3 = 10.48 (\pm 0.01) - 0.0615 (\pm 0.0004) \times \text{temp}$ .

Thus we can measure trends in Sr:Ca ratios in coral aragonite skeletons as a way to determine surface sea water temps.

# Ocean surface temps and climate in the Tertiary, determined by Sr ratios



Surface temps measured in shell-bearing sediment from the western Pacific Ocean. Relatively warm surface waters from 49-35 M.a. are followed by a climate change to cooling and the buildup of glaciers. By 2.5 M.a. the onset of glaciation in the Northern hemisphere occurred, several Ma after it started in Antarctica.

# Using a variety of data sets we have determined paleoclimates for the history of the Earth

- $O_{18}/O_{16}$  ratios; Sr:Ca ratios,  $Sr_{87}/Sr_{86}$
- Rock types (coal, tillites, evaporites, reefs)
- Vegetation types (pollen, leaf characteristics, tree rings)
- modelling

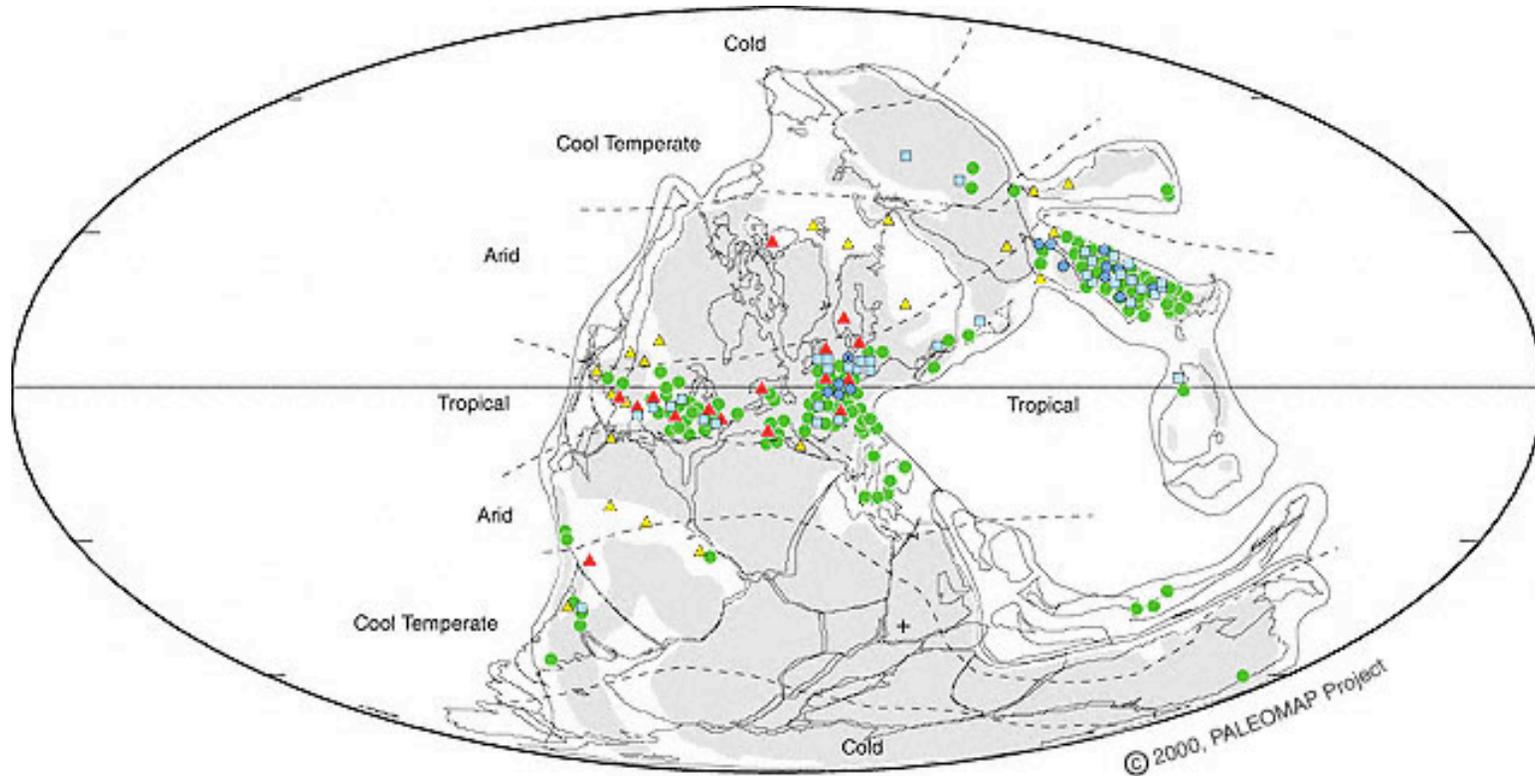
# “Greenhouse Earth” periods:

- Late Paleocene-Early Eocene (60-40Ma) - warm climates extend further north than today: England a tropical jungle! Palm trees in Alaska
- Late Cretaceous (100Ma-65Ma) - ocean temps rose  $> 5^{\circ}$ ; tropics widespread
- Triassic/Jurassic (250-140Ma) - warm climates, even at the poles

# “Icebox Earth” periods

- Late Pliocene-Pleistocene (5Ma-present) - we are still in a global cool period
- mid-late Carboniferous (320-290Ma)- glaciers in southern hemisphere up to 45°
- Late Ordovician (460-440Ma) - glaciers over continents N & S of equator
- Late Precambrian (1Ba-550 Ma) - “Snowball Earth”.....glaciers at the equator

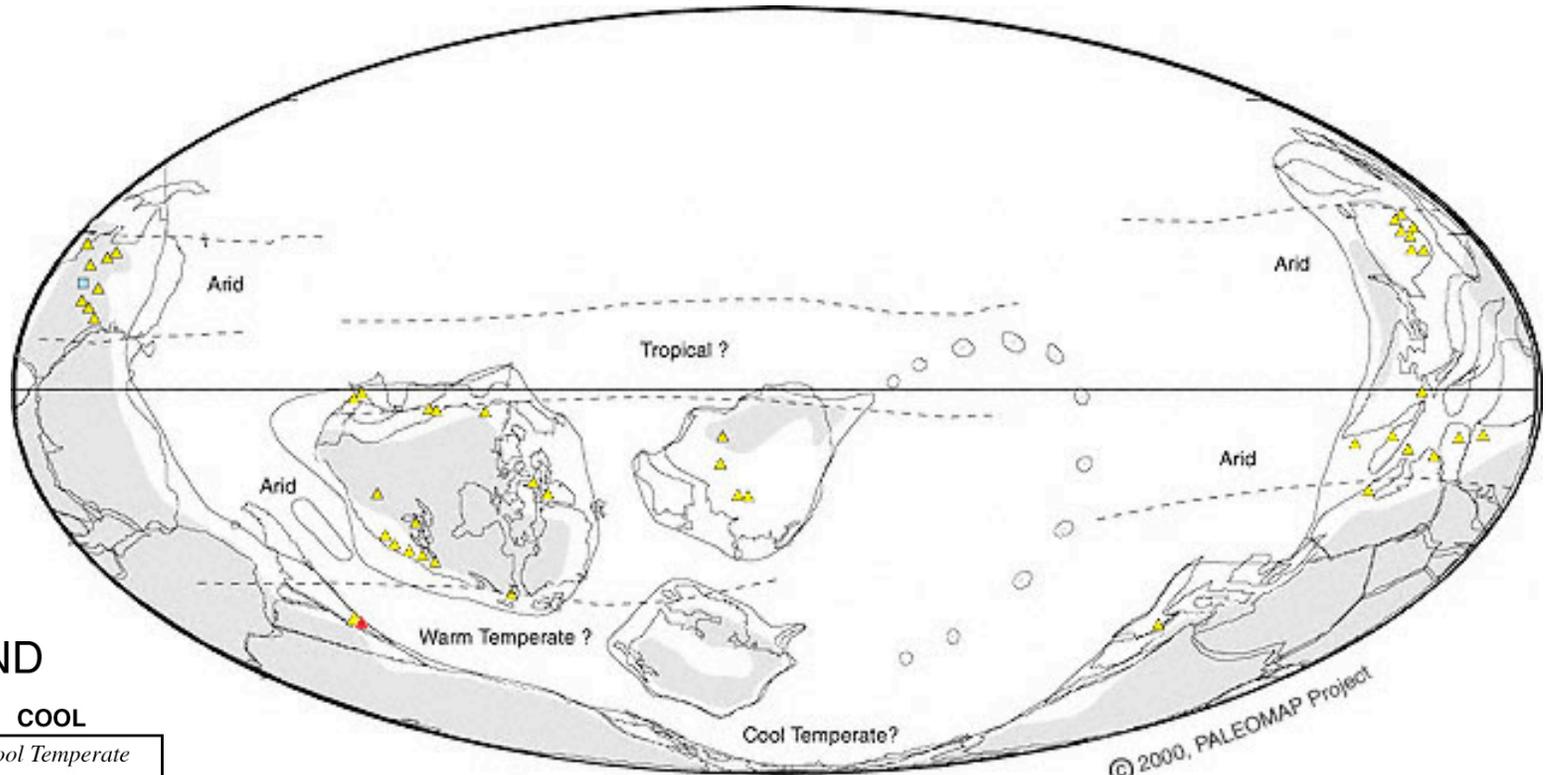
In addition to overall heating and cooling  
latitudinal climate zones can be  
compressed (not spaced out like today)



Upper Carboniferous (Gzelian)

Zonal compression can result from changes in the movement of heat from the equator to poles

# Hypothesize about the Cambrian climate of VT



## LEGEND

	WARM	COOL
	<i>Tropical</i>	<i>Cool Temperate</i>
	● Coal	● Coal & Tillites
	● Bauxite	
	● Laterite	
	<i>Warm Temperate</i>	
	□ Kaolinite (& coal & evaporite)	
	🌴 Crocodiles 🐊	
	🌴 Palms & Mangroves 🌴	
	<i>Arid</i>	<i>Cold</i>
WET	▲ Evaporite	⊕ Tillite
	▲ Calcrete	⊕ Dropstone
DRY		● Glendonite

Middle & Upper Cambrian

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"Paratropical" = High Latitude Bauxites

# Take away points..

- What is a “climate proxy?”
- What are examples of:  
geochemical proxies?  
biological proxies?  
lithologic proxies?

When you see how paleoclimate reconstruction for a geologic period, can you describe the various techniques used to create it?