# CHEM 023/025: OUTLINE OF GENERAL CHEMISTRY

Fall 2019

LECTURE A: CHEM 023 (90038) & CHEM 025 (90591), M,W,F 8:30AM-9:20AM, Innovation E102

**LECTURE B**: CHEM 023 (90788) & CHEM 025 (90592), T,Th 8:30AM-9:45AM, Marsh Life Sci 235

**GENERAL INFORMATION:** (see also the CHEM23 BlackBoard page)

**Instructor**: Steve Flemer **Email**: sflemer@uvm.edu

Office: 331 Innovation Office Hours: M W F 9:30 AM - 10:30 AM

T Th 10:00 AM – 11:00 AM

**Lecture**: The lecture will primarily be used to cover new material. Included in this syllabus is a tentative schedule covering the text material and the corresponding problems to be worked from each chapter.

**Exams**: Three 2-hour exams are given on Thursday nights from 6:00-8:00 PM.

|            | <b>Lecture A</b> (M,W,F; 8:30-9:20 AM)     | <b>Lecture B</b> (T,Th; 8:30-9:45 AM)           |
|------------|--|---|
| Exam 1     | Thurs, Sept. 19; Billings Lecture Hall     | Thurs, Sept. 19; 301 Williams                   |
| Exam 2     | Thurs, Oct. 17; Billings Lecture Hall      | Thurs, Oct. 17; 301 Williams                    |
| Exam 3     | Thurs, Nov. 14; Billings Lecture Hall      | Thurs, Nov. 14; 301 Williams                    |
| Final Exam | Mon, Dec. 9; 7:30-10:15AM; Innovation E102 | Tues, Dec. 10; 7:30-10:15AM; Marsh Life Sci 235 |

Absences from exams: Students with legitimate excuses (ie: a UVM-related conflict) may be permitted to take an exam sometime during the day that it is given to the rest of the class that evening. This must be cleared with the instructor first, however. Makeup exams will only be administered after the scheduled exam time if a medical or family emergency precludes taking the exam at the scheduled time.

**Review Sessions**: I will be holding Exam Review Sessions the Wednesday evenings prior to impending exams. Weekly SI sessions will also be starting shortly after the beginning of classes. Firm dates for these Review Sessions and SI sessions will be announced.

**Problems**: Exam questions will be modeled very closely to the type of problems you will encounter in the Practice Problems of each unit of study posted on BlackBoard. Solutions to all of these problems are included in these documents. While it is strongly suggested that you do as many problems as possible, the problems are not collected or graded.

Weekly Blackboard Quizzes: Each week, you will be responsible for taking a short online BlackBoard quiz covering the class material from the current week. Just click on the "Weekly Quiz" link on the left hand side of the CHEM23 BlackBoard page and follow the instructions. These quizzes are open-book, but must be completed independently. Weekly quizzes will be available to take until 5:00PM of the Sunday prior to a new week of classes. A skipped or a missed quiz is given a zero.

#### **REQUIRED COURSE MATERIALS:**

**Text**: There is <u>no textbook</u> for the course. Each unit of study has a corresponding folder in the Course Materials section of the course's BlackBoard site, within which are educational notes for that unit. These notes, while helpful for following along with the material, should not be thought of as comprehensive. Your own written class notes should be the basic core of your study materials.

Scientific Calculator: A standard scientific calculator is a requirement for the exams.

Note: Graphing calculators are not allowed.

Lab Manual: Available for download from the class' BlackBoard site.

Bound Laboratory Notebook: Available at the UVM Bookstore. Required for recording data.

(Note: the last two items are not required for CHEM 25 students).

#### **LABORATORY:**

(labs start 2 weeks after classes begin)

Time and Room: See your class course schedule as to your assignments.

Attendance: Students must attend the lab section they are assigned to. Official documentation of sickness or family crisis is required if a lab is missed. If more than 2 labs are missed, this results in a <u>failure</u> for the course. In order to take a lab at a time other than your assigned time one must obtain the permission of the TA and instructor.

Online Lab Safety Quiz: Prior to the lab sessions beginning, students must read through Lab Safety documentation and take a one-time online quiz before being allowed into their lab session. Just click the "Lab Safety" link on the left hand side of the CHEM23 BlackBoard page and follow the instructions. Students must score an 80 or better on the quiz to be admitted to lab. If you choose, you may take the Lab Safety quiz as many times as you want in order to maximize this score, as it will also count as your first lab quiz grade.

**Safety Eyewear**: OSHA approved safety glasses or goggles (available from the first floor stockroom or at the UVM Bookstore) must be worn by everyone once any experimentation has started in any area of a lab room.

**Foot Wear:** Only shoes that cover the toes are permitted in the lab. Sandals and open-toed shoes are not permitted.

### **ACADEMIC INTEGRITY:**

Offenses against the Code of Academic Integrity (ie: Cheating) are deemed serious and insult the integrity of the entire academic community. Any suspected violations of the code are taken very seriously and will be forwarded to the Center for Student Ethics & Standards for further investigation.

#### **COURSE GRADE FOR CHEM 23 STUDENTS:**

1. Points needed to obtain a specific grade

$$920 = A$$
  $870 = B+$   $790 = B 680 = C$   $620 = D+$   $570 = D 900 = A 820 = B$   $760 = C+$   $650 = C 590 = D$  less than  $570 = F$ 

2. How to calculate your points:

I will drop your lowest score. If the final exam is your lowest grade it will only count once. If your quiz average is your lowest grade, this score will be your drop. The 1.6 factor is because each test was only worth 100 pts, and therefore the maximum number of points obtainable from the tests are 500. In order to raise this to 800 pts you must multiply the  $500 \times 1.6 = 800$ .

#### Example:

| Actual Scores  | Ex-1<br>85 | Ex-2<br>45 | Ex-3<br>78 | Quiz Av.<br>77 | Final x 2 75 75 |
|----------------|------------|------------|------------|----------------|-----------------|
| Scores Counted | 85         | 75         | 78         | 77             | 75              |

Total pts =  $390 \times 1.6 = 624$  pts from class

b) Laboratory = 200 pts

|                   | 200 pts       |
|-------------------|---------------|
| Technique         | <u>25</u> pts |
| Quizzes           | 65 pts        |
| Lab reports       | 80 pts        |
| Notebook / Prelab | 30 pts        |

3. <u>Determination of grade</u>: Add up your points from the class and lab and then use the chart at the beginning to determine your course grade.

Example: 624 class pts + 160 lab pts = 784 total pts = C+

## **COURSE GRADE FOR CHEM 25 STUDENTS:**

Since there is no laboratory component to your grade, you will be graded on your exam/quiz scores exclusively. Your 5 highest scores will be multiplied by 2 (rather than 1.6).

# **LABORATORY SCHEDULE**

| <u>Date</u>     | Experiment Description   |
|-----------------|--|
| 9 – 12 SEPT     | <u>CHECK-IN</u> & Densities of Common Substances                   |
| 16 - 19 SEPT    | Determination of Heat Capacity Using Calorimetry                   |
| 23 - 26 SEPT    | Qualitative Analysis   |
| 30 SEPT – 3 OCT | Synthesis of Ionic Compound Alum from Aluminum Metal               |
| 7 - 10 OCT      | NO LABS (Fall Recess on Monday)                                    |
| 14 - 17 OCT     | Determination of a Compound's Empirical Formula                    |
| 21 - 24 OCT     | Reaction Stoichiometry & Equation Balancing                        |
| 28 - 31 OCT     | Determination of Limiting Reactant                                 |
| 4 - 7 NOV       | Determination of Acid Content in Pickle Juice using Titration      |
| 11 - 14 NOV     | Determination of Limestone Content in Soil using the Ideal Gas Law |
| 18 – 21 NOV     | NO LABS (Thanksgiving Break)                                       |
| 25 – 28 NOV     | Acid-Base Equilibria and Buffers                                   |
| 2 – 5 DEC       | LAB CHECKOUT   |

# TENTATIVE LECTURE SCHEDULE

UNIT 1 (Measurement & Problem Solving)

UNIT 2 (Matter & Energy)

UNIT 3 (Atoms & Elements)

Exam 1 (Thursday, Sept. 19; 6:00-8:00PM)

UNIT 4 (Electrons in Atoms)

UNIT 5 (Chemical Bonding)

UNIT 6 (Molecules & Compounds)

UNIT 7 (Chemical Composition)

Exam 2 (Thursday, Oct 17; 6:00-8:00PM)

UNIT 8 (Chemical Reactions)

**UNIT 9** (Quantities in Chemical Reactions)

UNIT 10 (Solutions)

UNIT 11 (Gases)

UNIT 12 (Liquids, Solids, & Intermolecular Forces)

Exam 3 (Thursday, Nov 14; 6:00-8:00PM)

UNIT 13 (Acids & Bases)

UNIT 14 (Chemical Equilibrium)

**Final Exam (Cumulative)**