1. GENERAL INFORMATION:

A. LECTURE:


Office Hours: T-Th 10:00-11:30, 1:30-3:30 and 5:30-6:15
Wed: 3:00-6:00 or by appointment

Lecture:
Lectures cover the material in the order assigned at the end of this syllabus in the tentative lecture section. Attendance at lecture is not mandatory. If you do come please stay through the lecture and refrain from talking to other students.

Exams: Exams are schedule for Wed nights from 6:15 - 9:00 pm in MLS 235. There are no make up dates. As only 3 out of the four exams given are used in determining your grade. A missed exam is the grade that is dropped.

While taking an exam only non-programable calculators are permitted. No other electronic devices are allowed (i.e. no cell phones, mp-3 players, ipods etc). Students caught using any electronic device other then a non-programable calculator will receive a zero for the exam. If you do wish to use a calculator during an exam it is the responsibility of the student to bring a non-programable calculator to the exam.

Listed below are the scheduled exam dates

Ex-1  Wed,  5 FEB  Ch: 4(sec .4-.7), all 12, 13
Ex-2  Wed, 12 MAR  Ch: sec 4.8, all of 14, 15
Ex-3  Wed,  2 APR  Ch: all 16 ,17
Ex-4  Wed, 23 APR  Ch: sec 4.9, all of 18, 19
Final  Tues,  6 MAY  Comprehensive (4:30-7:15 in Angell B 106)

Recitation: Wed, 6:15-7:15 in MLS 235

Review Sessions: week of exam only,  Sun 4:30 pm Angell B112,

Problems: Answers to the review questions and exercises are in the solutions manual. While it is strongly recommended that you do as many problems as possible, the problems are not collected and do not count towards your grade.
B. SUGGESTED READING:
Text: "Chemistry, A Molecular Approach" by Nivaldo Tro, volume 2, custom edition for UVM, is sold at the UVM bookstore. This is actually the third edition of Tro, but is the second time it has been packaged specially for UVM. The solutions manual that comes with the text has the complete solutions to all the assigned problems.

Login: jsharp@uvm.edu  Pass Word: SHARPSPRING2014

Lab Manual:
"Chemistry 32, A Lab Manual", sold at first floor stockroom, A-143 Cook, for $15.00

Notes & Old Tests:
Available on Blackboard

C. LABORATORY:

TA: TBA

Time:
As scheduled

Lab Videos:
Prior to the attending a lab session it is mandatory to watch the video that accompanies the lab. The videos demonstrate the proper use of new equipment, and the safe handling of chemicals. The videos are found at http://www.uvm.edu/~chem/courses/?Page=32Videos.html

Attendance:
Students must attend the lab section they are assigned to. Official documentation of sickness or family crisis is required if a lab is missed. Unexcused absences will result in a ZERO grade for that laboratory experiment. In order to take a lab at a time other then your assigned time you must obtain the permission of the lab coordinator, Christine Cardillo. If possible, please reschedule labs a week in advance. If for any reason more than two labs are missed you will receive an F for the course unless the dean of the college you are enrolled in grants you an incomplete.

Breakage Card:
A breakage card ( $40.00 ) must be purchased from the first floor stockroom, A-143 Cook, prior to your first lab. It is advisable to purchase this card as soon as possible in order to avoid having to wait in a line. The $40.00 is refundable if you are careful and avoid breaking your equipment. Remember don't leave home without it as it is required in order to be admitted into lab.
Safety eye wear:
OSHA approved safety glasses or goggles must be worn by everyone once any experimentation has started in any area of a lab room. Students not observing this rule will receive a ZERO for that experiment, warnings will not be given. Safety eye wear can be purchased at the UVM bookstore.

CONTACT LENSES are a potential hazard and should only be worn in the laboratory if you have no other type of corrective lenses. If you wear contact lenses you must wear goggles, and be sure to inform your TA that you are wearing contacts.

Foot Wear: Only shoes that cover the toes are permitted in the lab. Sandals, Crocs and other open toed shoes are not permitted.

Lab Notebook: A notebook with carbon-less copies must used to record data taken in the lab. All data must be recorded in ink at the time the data is generated.

D. COURSE GRADE:

Total Points = 1000  (800 class + 200 lab)

1. Points needed to obtain a specific grade

\[
\begin{align*}
A & \geq 910 \\
B+ & \geq 860 \\
B- & \geq 780 \\
C & \geq 650 \\
D+ & \geq 600 \\
D- & \geq 540 \\
A- & \geq 890 \\
B & \geq 800 \\
C+ & \geq 740 \\
C- & \geq 630 \\
D & \geq 560 \\
F & \leq 539
\end{align*}
\]

2. How to calculate your points

a) Class = 800pts

4 hr Exams = 4 grades
1 Final = 2 grades

6 grades - 1 grade = 5 grades x 1.6 = class pts

Only five grades are counted. If the final is your lowest grade it only counts once. If an hour exam is your lowest grade then one of the grades for the final will replace that low exam grade. The 1.6 factor is because each test was only worth 100 pts, and therefore the maximum number of points obtainable from the tests are 500. In order to raise this to 800 pts you must multiply the 500 x 1.6 = 800.

An option to taking the final:

You may elect not to take the final and use your lowest hourly exam score as the score for the final. You must let me know before you take the final exam if you wish to exercise this option. Once you start taking the final exam it will be counted.
b) **Laboratory = 200 pts**

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<thead>
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<tbody>
<tr>
<td></td>
<td>Notebook / Prelab</td>
<td>18 pts</td>
<td></td>
<td></td>
<td></td>
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<tr>
<td></td>
<td>Lab reports</td>
<td>110 pts</td>
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<td></td>
<td>Quizzes</td>
<td><strong>72 pts</strong></td>
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<td><strong>200 pts</strong></td>
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Obtained from lab TA, average grade is normally an 80% or 160 pts

**Absences:**

Unexcused absences from exams will result in a ZERO as a grade on your record. With that said, please contact me any time you have trouble making an exam to discuss possible make up options.

**Example 1:**

<table>
<thead>
<tr>
<th></th>
<th>Ex-1</th>
<th>Ex-2</th>
<th>Ex-3</th>
<th>Ex-4</th>
<th>Final x 2</th>
</tr>
</thead>
<tbody>
<tr>
<td>Actual Scores</td>
<td>85</td>
<td>45</td>
<td>78</td>
<td>77</td>
<td>80</td>
</tr>
<tr>
<td>Scores Counted</td>
<td>85</td>
<td>80</td>
<td>78</td>
<td>77</td>
<td>80</td>
</tr>
</tbody>
</table>

Total pts = 390 x 1.6 = 624 pts from class

**Example 2:**

<table>
<thead>
<tr>
<th></th>
<th>Ex-1</th>
<th>Ex-2</th>
<th>Ex-3</th>
<th>Ex-4</th>
<th>Final x 2</th>
</tr>
</thead>
<tbody>
<tr>
<td>Actual Scores</td>
<td>67</td>
<td>78</td>
<td>76</td>
<td>69</td>
<td>62</td>
</tr>
<tr>
<td>Scores Counted</td>
<td>67</td>
<td>78</td>
<td>76</td>
<td>69</td>
<td>62</td>
</tr>
</tbody>
</table>

Total pts = 352 x 1.6 = 563 pts from class

3. **Determination of grade.**

Add up your points from the class and lab and then use the chart at the beginning to determine your course grade.

**Example 1:**

\[
\begin{align*}
640 \text{ class pts} + 160 \text{ lab pts} & = 800 \text{ total pts} = B \\
\end{align*}
\]

**Example 2:**

\[
\begin{align*}
563 \text{ class pts} + 160 \text{ lab pts} & = 723 \text{ total pts} = C \\
\end{align*}
\]
# 2. TENTATIVE LABORATORY SCHEDULE

<table>
<thead>
<tr>
<th>DATE</th>
<th>EXPERIMENT</th>
</tr>
</thead>
<tbody>
<tr>
<td>27 - 30 JAN</td>
<td>Molar Mass from Freezing Point Depression</td>
</tr>
<tr>
<td>3 - 8 FEB</td>
<td>Iodination of Cyclohexanone</td>
</tr>
<tr>
<td>10 - 13 FEB</td>
<td>Keq of FeSCN$^{+2}$</td>
</tr>
<tr>
<td>17 - 20 FEB</td>
<td>Presidents Day - No Lab</td>
</tr>
<tr>
<td>24 - 27 FEB</td>
<td>Acid Neut. Pot. Anti-Acids</td>
</tr>
<tr>
<td>3 - 6 MAR</td>
<td>Spring Break - No Lab</td>
</tr>
<tr>
<td>10 - 13 MAR</td>
<td>Acid-base Equilibria and Buffers</td>
</tr>
<tr>
<td>17 - 20 MAR</td>
<td>$K_w$ of Copper (II) tartrate</td>
</tr>
<tr>
<td>24 - 27 MAR</td>
<td>Thermodynamics of the Dissolution of Borax</td>
</tr>
<tr>
<td>31 MAR - 3 APR</td>
<td>Oxidizing Power of Bleaches</td>
</tr>
<tr>
<td>7 - 10 APR</td>
<td>Potentiometric Det. of Ka</td>
</tr>
<tr>
<td>14 - 17 APR</td>
<td>Electrolysis/Electroplating Check-out</td>
</tr>
</tbody>
</table>

# 3. TENTATIVE LECTURER SCHEDULE

<table>
<thead>
<tr>
<th>DATE</th>
<th>CHAPTER(section)</th>
<th>SUGGESTED PROBLEMS</th>
</tr>
</thead>
<tbody>
<tr>
<td>14 - 16 JAN</td>
<td>Review Sections</td>
<td>Ch 4 sec .4 - .7 (Molarity, Net Ionic Rxns, Titration) Ch 11 sec .3 - .4 (Intermolecular forces)</td>
</tr>
<tr>
<td></td>
<td>Ch 12 All</td>
<td>Ch12 - 6,8,10,12,13,14,18,21,25,31,33,35,37,39,42,43,47,49,51,54,57,59,61,63,70,71,73,75,78,80,83,86,89,92,93,96,99,101,106,108,112115</td>
</tr>
<tr>
<td>21 JAN</td>
<td>Finish 12</td>
<td>Ch13 - 3,6,9,12,14,19,23,25,27,30,33,39,41,43,45,47,51,53,55,57,59,61,64,67,72,75,79,81,85,87,90,94,97,104,108</td>
</tr>
<tr>
<td></td>
<td>Start 13</td>
<td></td>
</tr>
<tr>
<td>23 JAN - 4 FEB</td>
<td>Finish 13</td>
<td></td>
</tr>
</tbody>
</table>
5 FEB  TEST 1:  All of 12, 13

6 - 13 FEB  14 All  Ch 14- 4 - 20,21,23,27,29,31,32,33,35,37,40,43,46,48, 53,55,58,62,63,69,71,74,77,79,81,85,86,89

18 - 27 FEB  Review Section  Ch 4 sec 8 (Acid-Base Rxns)
Start 15, will cover theory presented in sec 15.1- 15.4 and 15.10 first

15 All  Ch 15 - 2-6,8-30,34,35,36,37,39,41,45,49,53,57,61,63,65, 67,69,71,73,75,79,82,85,87,89,91,93,95,97,104, 106,110,111,115,117,120,123,125,130,137,142,145

3 - 6 MAR   OFF - SPRING BREAK

11 MAR  Finish 15

12 MAR  TEST 2:  All of 14, 15

13 MAR  16 All  Ch 16 - 2-23,27,29,31,34,39,41,45,47,49,51,53,55,57,59, 63,65,67,71,74,76,80,82,83,87,89,91,94,97,99,100, 103,107,114,117,120,123,126,130,134,136

18 - 20 MAR  Finish 16
17 All  CH 17 - 4-26,27,31,33,37,39,41,44,47,49,51,55,57,59,61, 63,65,67,71,73,75,77,79,82,85,87,88,89,93,98

25 - 1 APR  Finish 17

2 APR  TEST 3:  All of 16 ,17

3 - 10 APR  Review Section  Ch4 sec .9 (Acid-Base rxns)
18 All  Ch 18 - 4-36,39,41,43,45,47,49,51,53,57,59,61,63,65,67, 71,73,75,77,79,82,86,88,93,96,99,102, 103,109,113,117,121,123,125


23 APR  TEST 4:  All of 18, 19

24 - 29 APR  Review

6 MAY  Final - Cumulative  4:30-7:15 in Angell B106