

# CHEM 23/25: OUTLINE OF GENERAL CHEMISTRY

Spring 2016

**LECTURE:** CHEM 23 (10109) & CHEM 25 (10110), T,Th 8:30AM-9:45AM, Lafayette 207  
**RECITATION FOR BOTH CLASSES:** T 6:00-8:30 PM, Lafayette 207; LABS for CHEM 23 students only.

## GENERAL INFORMATION

**Instructor:** Dr. David Pratt

**Email:** dpratt1@uvm.edu

**Office:** A-109 Cook.

**Office Hours:** Mon. 9:00 – 11:30 AM; Tues. & Thurs. 10:00 – 11:30 AM.

**Class Website:** <https://Bb.uvm.edu>

**Lab Videos:** <http://www.uvm.edu/~chem/?Page=23Videos.html>

**Lecture:** The lecture will provide an overview of all material to be discussed in this course. Key topics include the chemical world, measurement and problem solving, matter and energy, the chemistry of the elements and their compounds, and the concepts of chemical bonding, chemical kinetics, and chemical equilibrium. Brief introductions to the topics of organic chemistry, biochemistry, and nuclear chemistry will be included.

## REQUIRED TEXTBOOKS

**Text:** "Introductory Chemistry", 5th edition, by Nivaldo J. Tro, sold at the UVM bookstore.

**Mastering Chemistry:** An on-line homework and tutorial system, also available at the bookstore.

**Lab Manuals:** "Chemistry 23 Experiments " is available at the class Blackboard site, free of charge. **(Not required for CHEM 25 students).**

**Scientific Calculator:** A standard scientific calculator is a requirement for the exams.

**Note:** Graphing calculators are not allowed.

## CLASS ACTIVITIES

Pre-lecture questions, lecture participation, homework, office hours, recitations, labs, and exams.

## CLASS SCHEDULE

Date	Topic	Homework Due
January 19*	The Chemical World (Chapter 1)	January 26 (#1)
January 21	Measurement and Problem Solving (2)	January 26(#1)
January 26*, 28	Matter and Energy (3)	February 1(#2)
February 2*	Atoms and Elements (4)	February 9(#3)
February 4	Molecules and Compounds (5)	February 9(#3)
February 9*, 11	Chemical Composition (6)	February 16(#4)
<b>February 16*</b>	Review (AM) and Exam (1-6) (PM)	---
February 18	Chemical Reactions (7)	February 23(#5)
February 23*, 25	Quantities in Chemical Reactions (8)	March 1(#6)
March 1*, 3	The Periodic Table (9)	March 15(#7)
March 8,10	Spring Break	---
March 15*, 17	Chemical Bonding (10)	March 22(#8)
<b>March 22*</b>	Review (AM) and Exam (7-10) (PM)	---
March 24	Gases (11)	March 29(#9)
March 29*	Liquids, Solids, and Intermolecular Forces (12)	April 5(#10)
March 31	Solutions (13)	April 5(#10)
April 5*,7	Acids and Bases (14)	April 12(#11)
April 12*	Chemical Equilibrium (15)	April 19 (#12)
April 14	Oxidation and Reduction (16)	April 19 (#12)
<b>April 19*</b>	Review (AM) and Exam (11-16) (PM)	---
April 21	Nuclear Chemistry (17)	April 26 (#13)
April 26*, 28	Organic Chemistry (18)	May 3 (#14)
May 3*	Biochemistry (19)	---
Date to be determined	<b>Final Exam (Chapters 1-19)</b>	Rooms to be determined.

**Problems:** Weekly problem sets will be assigned and graded on Mastering. Solutions to the assigned problems will be discussed in the evening recitations, marked with asterisks.

**Review Sessions:** Exam review sessions will be scheduled on the lecture day immediately preceding the exam to be given that evening.

**Absences from exams:** Students with legitimate excuses (*e.g.*, religious holidays, UVM-related conflicts, or family emergencies) will be permitted to take early exams providing they obtain permission from Dr. Pratt at least one week in advance of the scheduled exam time. Makeup exams, if permitted, will not be given *after* the scheduled exam time.

## GRADING

Final grades will be based on the percentage of total available points received. The available points include 100 for pre-lecture and office hour participation, 100 for homework, 200 for lab (Chem 23 students only), 200 for hour exams (best 2 of 3), and 200 points for the final. The percentages of points needed to obtain a specific grade are as follows:

**A (90% or higher), B (75%), C (60%), D (50%) and F (49% or lower).**

# MasteringChemistry®

## Student Registration

In this course you will be using MasteringChemistry®, an online tutorial and homework program that accompanies your textbook. *If you have joined a MasteringChemistry course before and can still log in:*

Save time by following the guide for joining another course found under the STUDENT heading at [www.masteringchemistry.com](http://www.masteringchemistry.com) > *Tours & Training*> *Getting Started* instead of using the steps below.

### What You Need:

- ✓ **A valid email address**
- ✓ **A student access code**  
(Comes in the Student Access Code Card/Kit that may have been packaged with your new textbook or that may be available separately in your school's bookstore. Otherwise, you can purchase access online at [www.masteringchemistry.com](http://www.masteringchemistry.com).)
- ✓ **The ZIP or other postal code for your school: 05405**
- ✓ **A Course ID: PRATT2016**

### 1. Register

- Go to [www.masteringchemistry.com](http://www.masteringchemistry.com) and click **Students** under **Register**.
- To register using the student access code inside the MasteringChemistry Student Access Code Card/Kit, select **Yes, I have an access code**. Click **Continue**.

–OR– *Purchase access online:* Select **No, I need to purchase access online now**. Select your textbook, whether you want access to the eText, and click **Continue**. Follow the on-screen instructions to purchase access using a credit card. The purchase path includes registration, but the process is a bit different from the steps printed here.

- **License Agreement and Privacy Policy:** Click **I Accept** to indicate that you have read and agree to the license agreement and privacy policy.
- Select the appropriate option under “Do you have a Pearson Education account?” Continue to give the requested information until you complete the process. The **Confirmation & Summary** page confirms your registration. This information will also be emailed to you for your records. You can either click **Log In Now** or return to [www.masteringchemistry.com](http://www.masteringchemistry.com) later.

### 2. Log In

- Go to [www.masteringchemistry.com](http://www.masteringchemistry.com).
- Enter your Login Name and Password that you specified during registration and click **Log In**.

### 3. Join Your Instructor's Online Course and/or Open Self-Study Resources

Upon first login, you'll be asked to do one or more of the following:

- **Join a Course** by entering the **MasteringChemistry Course ID** provided by your instructor. If you don't have a Course ID now, you can return to join the MasteringChemistry course later. When you join a course, you may also be asked for a Student ID (follow on-screen instructions).
- **Explore the Study Area** or **Launch Your eText**, if these resources are available for your textbook.

### To Access MasteringChemistry Again Later

Simply go to [www.masteringchemistry.com](http://www.masteringchemistry.com), enter your Login Name and Password, and click **Log In**.

*After you have joined a course:* You can open any assignments from the **Assignments Due Soon** area or from the **Assignments** page. For self-study, click **eText** or **Study Area**, if these options are available.

### Support

Access Customer Support at <http://www.masteringchemistry.com/support>, where you will find:

- System Requirements
- Answers to Frequently Asked Questions
- Registration Tips & Tricks video
- Additional contact information for Customer Support, including Live Chat

## **LABORATORY**

**Time and Room:** Labs begin on January 25. See your class course schedule regarding your assignments.

**Attendance:** Students must attend the lab section to which they are assigned. Official documentation of sickness or family crisis is required if a lab is missed. **If more than 2 labs are missed, this results in a failure for the course.** In order to take a lab at a time other than your assigned time one must obtain the permission of the TA and instructor.

**Breakage Card:** A breakage card (\$40.00) must be purchased from the first floor stockroom, A-143 Cook, prior to your first lab. The \$40.00 is refundable, and if you are careful you should get most of it back. Remember, you must have it with you to be admitted into lab.

**Safety Eyewear:** OSHA approved safety glasses or goggles must be worn by everyone once any experimentation has started in any area of a lab room. Safety eyewear can be purchased at the UVM bookstore.

**Foot Wear:** Only shoes that cover the toes are permitted in the lab. Sandals and open-toed shoes are not permitted.

**Lab Notebook:** A bound notebook is required for recording lab data.

## **ACADEMIC INTEGRITY**

Each student in this class is expected to be familiar with the UVM Code of Academic Integrity <http://www.uvm.edu/policies/student/acadintegrity.pdf> The principal objective of this code is to promote an intellectual climate that is consistent with and promotes the goals of a higher education. Offenses against this code in the lectures, labs, and/or exams, and on homework, will be deemed serious and will be reported to the Center for Student Ethics & Standards for further investigation. These offenses include copying homework, plagiarism, sharing results with other students in the lab, falsifying lab reports, and cheating on exams. If you have any concerns that a standard in this code may have been violated, you are expected to report it to Dr. Pratt or to Dr. Cardillo immediately.

## CHEMISTRY 23 LABORATORY SCHEDULE SPRING 2016.

Week	Dates	Labs	Work Due
1	Mon Jan 18 – Fri Jan 22	<b>No Labs</b> Purchase your Breakage Card and Safety Glasses; Review the Online Safety Presentation and complete the Safety Quiz <b>BEFORE</b> your first lab.	
2	Mon Jan 25 – Wed Jan 27	Safety Quiz Due, Laboratory <b>Check-In</b> , & <b>Experiment 1:</b> Determination of the Densities of Common Substances	Safety Quiz
3	Mon Feb 1 – Wed Feb 3	<b>Experiment 2:</b> Determination of Heat Capacity Using Calorimetry	Exp 1 Lab Report Exp 2 Pre-lab Exp 2 Quiz
4	Mon Feb 8 – Wed Feb 10	<b>Experiment 3:</b> Synthesis of the Ionic Compound Alum from Aluminum Metal	Exp 2 Lab Report Exp 3 Pre-lab Exp 3 Quiz
5	Mon Feb 15 – Wed Feb 17	<b>Presidents' Day Holiday</b> <b>No lab this week</b>	
6	Mon Feb 22 – Wed Feb 24	<b>Experiment 4:</b> Determination of a Compound's Empirical Formula	Exp 3 Lab Report Exp 4 Pre-lab Exp 4 Quiz
7	Mon Feb 29 – Wed Mar 2	<b>Town Meeting Day Recess</b> <b>No lab this week</b>	
<i>Spring Break</i>	<i>Mon Mar 7 – Fri Mar 11</i>	<i>Spring Break</i>	
8	Mon Mar 14 – Wed Mar 16	<b>Experiment 5:</b> Reaction Stoichiometry and Equation Balancing	Exp 4 Lab Report Exp 5 Pre-lab Exp 5 Quiz
9	Mon Mar 21 – Wed Mar 23	<b>Experiment 6:</b> Determination of Limiting Reactant	Exp 5 Lab Report Exp 6 Pre-lab Exp 6 Quiz
10	Mon Mar 28 – Wed Mar 30	<b>Experiment 7:</b> Determination of Limestone Content in Soil Using the Ideal Gas Law	Exp 6 Lab Report Exp 7 Pre-lab Exp 7 Quiz
11	Mon Apr 4 – Wed Apr 6	<b>Experiment 8:</b> Qualitative Analysis	Exp 7 Lab Report Exp 8 Pre-lab Exp 8 Quiz
12	Mon Apr 11 – Wed Apr 13	<b>Experiment 9:</b> Determination of Acid Content in Pickle Juice Using Titration	Exp 8 Lab Report Exp 9 Pre-lab Exp 9 Quiz
13	Mon Apr 18 – Wed Apr 20	<b>Experiment 10:</b> Acid-Base Equilibria and Buffers	Exp 9 Lab Report Exp 10 Pre-lab Exp 10 Quiz
14	Mon Apr 25 – Wed Apr 27	Laboratory <b>Clean-up</b> & Laboratory <b>Check-Out</b> All work must be handed in at your last lab.	Exp 10 Lab Report
15	Mon May 2 – Wed May 4	<b>NO LABS</b>	
Finals	Fri May 6 – Fri May 13	Final Exams <i>Good Luck!</i>	

